

GCE Chemistry Data Sheet


Table 1
Infrared absorption data

Bond	Wavenumber /cm ⁻¹
N-H (amines)	3300–3500
O-H (alcohols)	3230–3550
C-H	2850–3300
O-H (acids)	2500–3000
C≡N	2220–2260
C=O	1680–1750
C=C	1620–1680
C-O	1000–1300
C-C	750–1100

Table 2
¹H n.m.r. chemical shift data

Type of proton	δ/ppm
ROH	0.5–5.0
RCH ₃	0.7–1.2
RNH ₂	1.0–4.5
R ₂ CH ₂	1.2–1.4
R ₃ CH	1.4–1.6
$\begin{array}{c} \\ \text{R}-\text{C}-\text{C}- \\ \quad \\ \text{O} \quad \text{H} \end{array}$	2.1–2.6
$\begin{array}{c} \text{R}-\text{O}-\text{C}- \\ \\ \text{H} \end{array}$	3.1–3.9
RCH ₂ Cl or Br	3.1–4.2
$\begin{array}{c} \\ \text{R}-\text{C}-\text{O}-\text{C}- \\ \quad \\ \text{O} \quad \text{H} \end{array}$	3.7–4.1
$\begin{array}{c} \text{R} \quad \text{H} \\ \diagdown \quad / \\ \text{C}=\text{C} \\ / \quad \diagdown \end{array}$	4.5–6.0
$\begin{array}{c} \text{O} \\ // \\ \text{R}-\text{C} \\ \\ \text{H} \end{array}$	9.0–10.0
$\begin{array}{c} \text{O} \\ // \\ \text{R}-\text{C} \\ \\ \text{O}-\text{H} \end{array}$	10.0–12.0

Table 3
¹³C n.m.r. chemical shift data

Type of carbon	δ/ppm
$\begin{array}{c} \quad \\ -\text{C}-\text{C}- \\ \quad \end{array}$	5–40
$\begin{array}{c} \\ \text{R}-\text{C}-\text{Cl or Br} \\ \end{array}$	10–70
$\begin{array}{c} \quad \\ \text{R}-\text{C}-\text{C}- \\ \quad \\ \text{O} \quad \end{array}$	20–50
$\begin{array}{c} \\ \text{R}-\text{C}-\text{N} \\ \quad \diagdown \end{array}$	25–60
$\begin{array}{c} \\ -\text{C}-\text{O}- \\ \end{array}$ alcohols, ethers or esters	50–90
$\begin{array}{c} \diagdown \quad / \\ \text{C}=\text{C} \\ / \quad \diagdown \end{array}$	90–150
R-C≡N	110–125
	110–160
$\begin{array}{c} \text{O} \\ // \\ \text{R}-\text{C}- \\ \end{array}$ esters or acids	160–185
$\begin{array}{c} \text{O} \\ // \\ \text{R}-\text{C}- \\ \end{array}$ aldehydes or ketones	190–220



The Periodic Table of the Elements

1		2												3	4	5	6	7	0				
																						(18)	
																						4.0 He helium 2	
																						20.2 Ne neon 10	
																						39.9 Ar argon 18	
																						83.8 Kr krypton 36	
																						131.3 Xe xenon 54	
																						[222] Rn radon 86	
																						[223] Fr francium 87	
																						[226] Ra radium 88	
																						[227] Ac † actinium 89	
																						[267] Rf rutherfordium 104	
																						[268] Db dubnium 105	
																						[271] Sg seaborgium 106	
																						[272] Bh bohrium 107	
																						[270] Hs hassium 108	
																						[276] Mt meitnerium 109	
																						[281] Ds darmstadtium 110	
																						[280] Rg roentgenium 111	
																						Elements with atomic numbers 112-116 have been reported but not fully authenticated	

Key
relative atomic mass
symbol
name
atomic (proton) number

1.0
H
hydrogen
1

* **58 – 71** Lanthanides

† **90 – 103** Actinides

140.1 Ce cerium 58	140.9 Pr praseodymium 59	144.2 Nd neodymium 60	[145] Pm promethium 61	150.4 Sm samarium 62	152.0 Eu europium 63	157.3 Gd gadolinium 64	158.9 Tb terbium 65	162.5 Dy dysprosium 66	164.9 Ho holmium 67	167.3 Er erbium 68	168.9 Tm thulium 69	173.1 Yb ytterbium 70	175.0 Lu lutetium 71
232.0 Th thorium 90	231.0 Pa protactinium 91	238.0 U uranium 92	[237] Np neptunium 93	[243] Pu plutonium 94	[243] Am americium 95	[247] Cm curium 96	[247] Bk berkelium 97	[251] Cf californium 98	[252] Es einsteinium 99	[257] Fm fermium 100	[258] Md mendelevium 101	[259] No nobelium 102	[262] Lr lawrencium 103