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I declare this is my own work.

GCSE COMBINED SCIENCE: SYNERGY

F

Foundation Tier Paper 4 Physical Sciences

8465/4F

Tuesday 13 June 2023 Morning

Time allowed: 1 hour 45 minutes

At the top of the page, write your surname and forename(s), your centre number, your candidate number and add your signature.



MATERIALS

For this paper you must have:

- a ruler
- a protractor
- a scientific calculator
- the periodic table (enclosed)
- the Physics Equations Sheet (enclosed).

INSTRUCTIONS

- Use black ink or black ball-point pen. Pencil should only be used for drawing.
- Answer ALL questions in the spaces provided. Do not write on blank pages.
- If you need extra space for your answer(s), use the lined pages at the end of this book. Write the question number against your answer(s).
- Do all rough work in this book. Cross through any work you do not want to be marked.
- In all calculations, show clearly how you work out your answer.



INFORMATION

- The maximum mark for this paper is 100.
- The marks for questions are shown in brackets.
- You are expected to use a calculator where appropriate.
- You are reminded of the need for good English and clear presentation in your answers.

DO NOT TURN OVER UNTIL TOLD TO DO SO



0 1
The stopping distance of a vehicle depends on the thinking distance and the braking distance.
0 1.1 What is meant by 'thinking distance'? [1 mark]
Tick (✓) ONE box.
The distance travelled before a vehicle stops.
The distance travelled while the driver reacts.
The time taken for a driver to react.
The time taken for the vehicle to stop.



0 1.2
What would increase the BRAKING DISTANCE of a vehicle? [1 mark]
Tick (✓) ONE box.
Ice on the road surface
Sunny weather
Using a mobile phone while driving
[Turn over]



0 1]. 3
	is the name of the force which causes the vehicle celerate when the brakes are applied? [1 mark]
Tick ((✓) ONE box.
	Friction
	Upthrust
	Weight



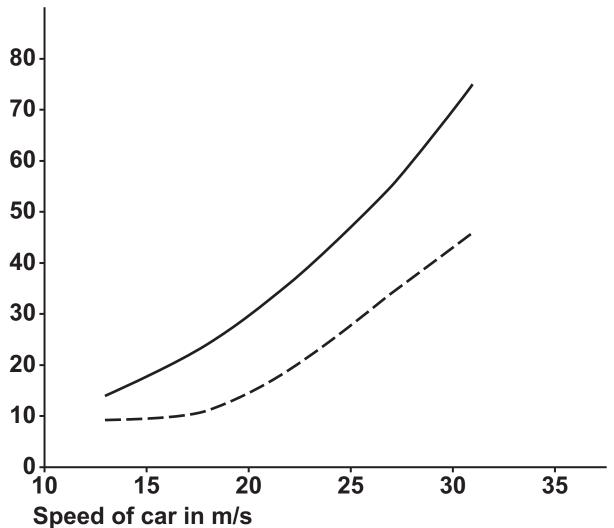
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FIGURE 1 shows how the braking distance of two cars varies with speed.

FIGURE 1

Braking distance in metres



KEY

Braking distance of car A

--- Braking distance of car B



0 1 . 4
How does the braking distance of the two cars vary with speed? [1 mark]
Tick (✓) ONE box.
The braking distance decreases as speed increases.
The braking distance is not affected by speed.
The braking distance increases as speed increases.
[Turn over]



To make a fair comparison	between the braking distance
of the two cars, the mass o	f each car was the same.

- - - -	0	1		5
---------------	---	---	--	---

Calculate the weight of a car.

mass = 850 kg

gravitational field strength = 9.8 N/kg [3 marks]

Use the equation:

weight = mass × gravitational field strength

Choose the unit from the the list below.

KILOGRAMS	METRES	NEWTONS

Weight = _____ Unit _____



0 1 . 6
Which TWO variables should be kept the same to make a fair comparison of the braking distance of the two cars? [2 marks]
Tick (✓) TWO boxes.
The age of the driver
The caffeine intake of the driver
The colour of the car
The number of people in the car
The type of road surface
[Turn over]



Mean braking force =	N
mean braking force = mass × deceleration [2 marks]	
Use the equation:	
Calculate the mean braking force on the car.	
At one speed the deceleration of one of the cars was 10.7 m/s ² .	
The mass of each car was 850 kg.	
0 1.7	



TABLE 1 shows the braking force on each car at a speed of 31 m/s.

TABLE 1

CAR	BRAKING FORCE IN N
A	5450
В	8880

The braking distance of car A was longer than the braking distance of car B at a speed of 31 m/s.

Explain why.

Use data from TABLE 1. [2 marks]						



0 2

This question is about Group 7 elements.

FIGURE 2 shows some Group 7 elements in the periodic table.

FIGURE 2

19
F
fluorine
9
35.5
Cl
chlorine
17
80
Br
bromine
35
127
I
iodine
53



0 2 . 1 What is the name of the Group 7 elements? [1 mark]
Tick (✓) ONE box.
Alkali metals
Halogens
Noble gases
0 2 . 2
Which element in FIGURE 2, on the opposite page, has the highest melting point? [1 mark]
[Turn over]



0	2		3
_	_	- 1	_

A fluorine atom has 9 electrons.

What is the electronic structure of a fluorine atom? [1 mark]

Tick (✓) ONE box.











0 2 . 4
What type of structure does chlorine have? [1 mark]
Tick (✓) ONE box.
Giant covalent
Ionic lattice
Small molecules
0 2 . 5
Name the compound produced when potassium reacts with iodine. [1 mark]
[Turn over]



A less reactive Group 7 element can be displaced from an aqueous solution of its salt by a more reactive Group 7 element.

Complete TABLE 2 to show if a reaction occurs when the element is added to the aqueous solution.

Use:

- ✓ where a reaction occurs
- * where there is no reaction. [3 marks]

TABLE 2

	AQUEOUS SOLUTION			
ELEMENT	Sodium fluoride	Sodium chloride	Sodium bromide	
Fluorine	×			
Chlorine		×		
Bromine			×	



8

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0 3

A student investigated resistance in a circuit.

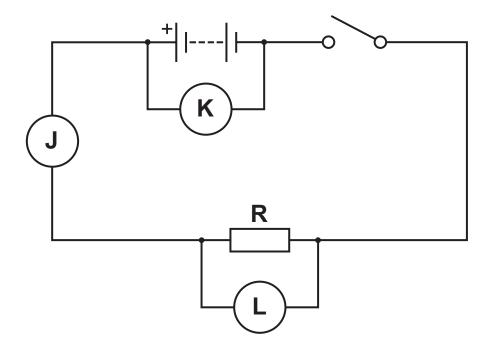
The student measured:

- the current in the resistor with an ammeter
- the potential difference across the resistor with a voltmeter.

0 3 . 1

FIGURE 3 shows a circuit diagram.

FIGURE 3





Which letter on FIGURE 3, on the opposite page,
shows the correct position for the ammeter to measure
the current in the resistor? [1 mark]

Tick (✓) ONE b	ox.		
J	K	L	
0 3 . 2 Draw the circui [1 mark]	t symbol for a vo	oltmeter in the box be	low



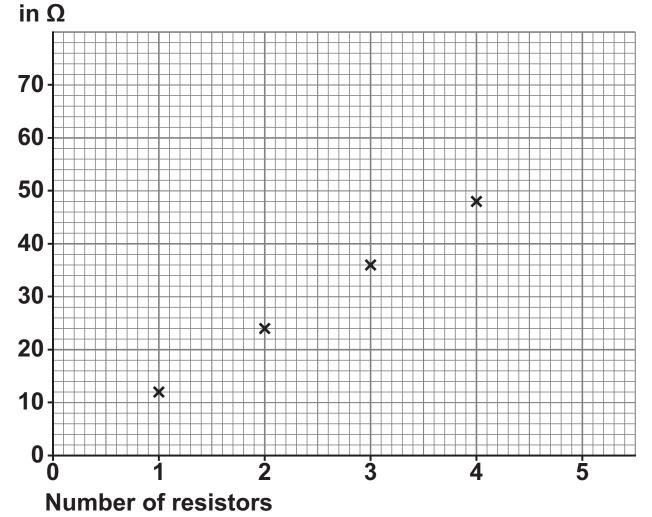
The student changed the number of identical resistors in the circuit.

The student calculated the total resistance of the resistors.

FIGURE 4 shows the results.

FIGURE 4

Total resistance





0 3 . 3	
Draw a line of best fit on FIGURE 4, on the opposite page. [1 mark]	
0 3 . 4	
Predict the total resistance when 5 resistors were connected in the circuit.	
You should extend your line of best fit. [1 mark]	
Total resistance =	_ Ω



0 3 . 5
How were the identical resistors connected in the student's circuit?
Use FIGURE 4, on page 22. [2 marks]
Tick (✓) ONE box.
The resistors were connected in parallel.
The resistors were connected in series.
The resistors were connected in series and in parallel.
Give a reason for your answer.



0 3.6	
The potential difference across the battery was 3.0 V.	
The maximum current in the circuit in the student's investigation was 0.25 A.	
Calculate the maximum power output of the battery in the student's investigation.	
Use the equation:	
power = potential difference × current [2 marks]	
Power = W	8
[Turn over]	



0 4
This question is about metal extraction and recycling.
Most metals are found in the Earth as metal compounds.
0 4 . 1
Why is gold found in the Earth as the metal itself? [1 mark]
0 4 . 2
How is aluminium extracted from aluminium compounds? [1 mark]
Tick (✓) ONE box.
By displacement
By electrolysis
By reduction with carbon



ı	Iron	ic	extracted	from	iron	ovido
ı	iron	IS	extracted	Trom	iron	oxiae.

0	4		3
		I - I	_

An oxide ion is an oxygen atom that has gained two electrons.

What is the formula of an oxide ion? [1 mark]

Tick (✓) ONE box.



0 4 . 4

Iron oxide is heated with carbon to produce iron and carbon dioxide.

The word equation for the reaction is:



Iron is used to make steel.
Scrap steel can be recycled.
0 4 . 5
Give TWO environmental advantages of recycling scrap steel instead of making new steel from iron. [2 marks]
1
2



0	4		6
---	---	--	---

TABLE 3 shows information about scrap steel in one year.

TABLE 3

	MASS IN MILLIONS OF KG
Scrap steel recycled	420
Total scrap steel	560

recycled in one year. [2 marks]	
Percentage =	



0 4 . 7

FIGURE 5 shows metal food and drink cans.

FIGURE 5



The metal cans are made from steel or aluminium.

Which property of steel means that steel can be separated from aluminium at a recycling centre? [1 mark]

J



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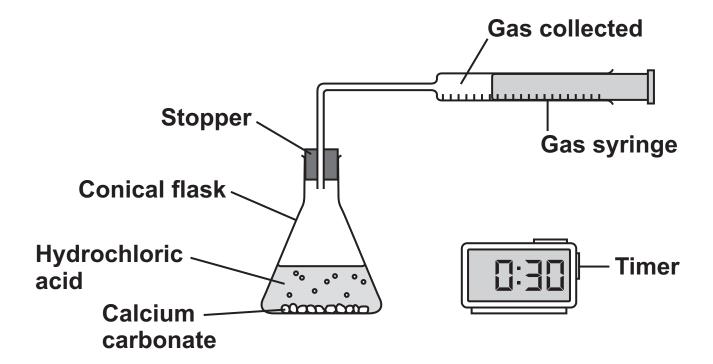
0 5

A student investigated the rate of the reaction between calcium carbonate and hydrochloric acid.

The word equation for the reaction is:

FIGURE 6 shows the apparatus.

FIGURE 6





This is the method used.

- 1. Add 50 cm³ of hydrochloric acid to the conical flask.
- 2. Add 2.0 g of calcium carbonate to the conical flask.
- 3. Immediately insert the stopper into the conical flask and start the timer.
- 4. Record the volume of gas collected in the gas syringe every 30 seconds.
- 5. Stop recording when the volume of gas in the gas syringe does not change.



Why must the stopper be inserted IMMEDIATELY after the reactants are added to the flask? [1 mark]
0 5 . 3
Name ONE piece of equipment that can be used to measure the mass of calcium carbonate. [1 mark]
Test tube
Measuring cylinder
Beaker
Tick (✓) ONE box.
Which piece of equipment is the most suitable to measure the volume of hydrochloric acid? [1 mark]
0 5 . 1



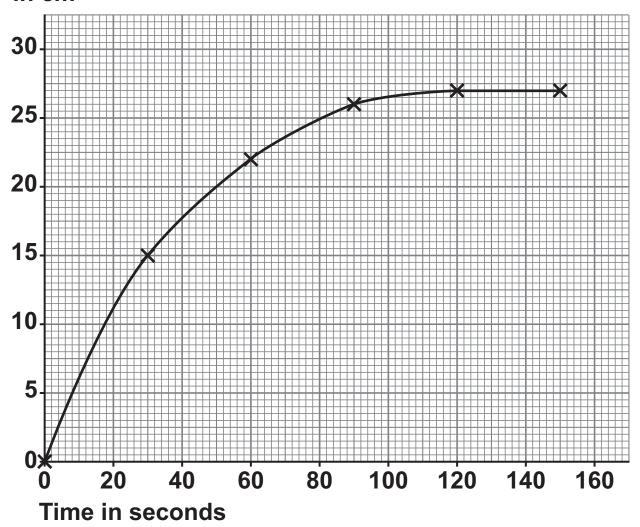
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FIGURE 7 shows the results.

FIGURE 7

Volume of gas collected in cm³





0	5	4
U	O	4

Describe the trend for the volume of gas collected.

Use FIGURE 7, on the opposite page. [2 marks]

Determine the mean rate of reaction from 0 to 30 seconds.

Use FIGURE 7.

Use the equation:

$$\label{eq:mean_rate} \text{mean rate of reaction} = \frac{\text{volume of gas collected}}{\text{time taken}}$$



Choose the u	nit from the list	below. [4 marks]	
cm ³ /s	s/cm ³	scm ³	
Mean rate of	reaction =		
Unit			



0 5 . 6	
---------	--

The student repeated the investigation using hydrochloric acid of higher concentration.

Complete the sentences.

Choose answers from the list below. [2 marks]

DECREASES STAYS THE SAME INCREASES

When the concentration of hydrochloric acid increases, the rate of reaction increases.

This is because the mean distance between particles

Therefore the frequency of collisions between particles



0 5.7	
What is the minimum amount of energy that particles must have to react? [1 mark]	
Tick (✓) ONE box.	
Activation energy	
Kinetic energy	
Potential energy	12



0 6
Permanent magnets have a magnetic field around them
0 6.1 The Earth has a magnetic field.
Which part of the Earth's internal structure creates the magnetic field? [1 mark]
Tick (✓) ONE box.
The crust
The mantle
The outer core
[Turn over]

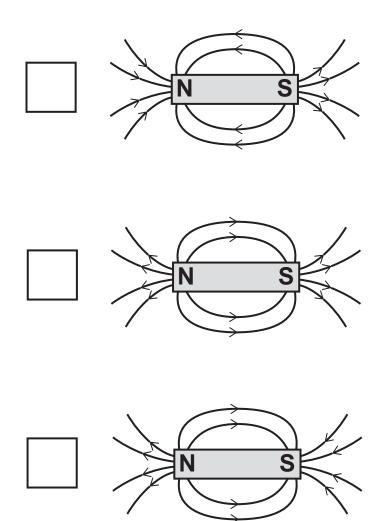




A magnetic compass contains a bar magnet.

Which diagram shows the magnetic field around a bar magnet? [1 mark]

Tick (✓) ONE box.

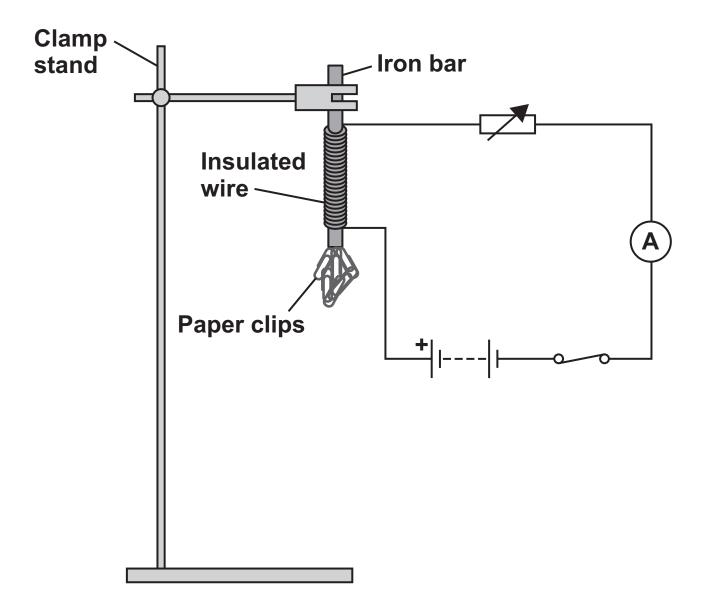




A student wrapped insulated wire around an iron bar to make an electromagnet.

FIGURE 8 shows the electromagnet held in a clamp and connected to a circuit.

FIGURE 8





When the switch was closed, the electromagnet attracted paper clips.

The student investigated how the number of turns of wire on the iron bar affected the strength of the electromagnet.

This is the method used.

- 1. Wrap 20 turns of wire around the iron bar.
- 2. Hold the paper clips near the bottom of the electromagnet.
- 3. Close the switch.
- 4. Count the number of paper clips the electromagnet picks up.
- 5. Open the switch.
- 6. Repeat steps 1 to 5, but with more turns of wire around the iron bar.



0 6.3
The student used the same value of current throughout the investigation.
What type of variable was the current? [1 mark]
Tick (✓) ONE box.
Control variable
Dependent variable
Independent variable



TABLE 4 shows the results.

When there was no wire wrapped around the iron bar, no paper clips were picked up.

TABLE 4

NUMBER OF TURNS OF WIRE	NUMBER OF PAPER CLIPS PICKED UP			
	TEST 1	TEST 2	TEST 3	MEAN
0	0	0	0	0
20	5	6	7	6
40	19	19	7	19
60	33	35	37	35
80	54	52	56	54

0 6.4

Draw a ring around the anomalous result in TABLE 4. [1 mark]



0 6 . 5

The student did NOT take another reading to replace the anomalous result.

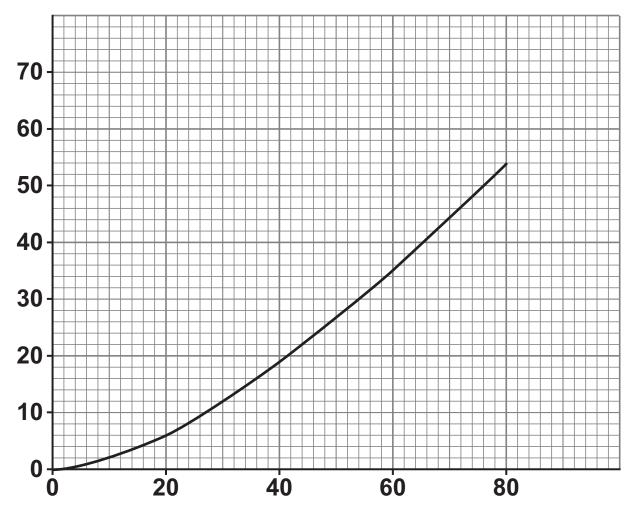
What did the student do with the anomalous result wher calculating the mean? [1 mark]				



FIGURE 9 is a graph of the results.

FIGURE 9

Mean number of paper clips picked up



Number of turns of wire



06.6	
How many paper clips would be picked up when 70 turns of wire are wrapped around the iron bar? [1 mark]	>

ound the iron bar? [1 mark]

Number of paper clips = _____

0 | 6 | . | 7 |

The investigation was repeated using a larger current.

Draw a line on FIGURE 9, on the opposite page, to show the expected results. [2 marks]



0 7

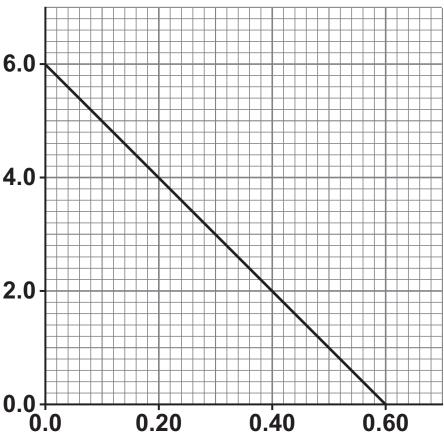
A student threw a ball vertically upwards into the air.

FIGURE 10 is a velocity-time graph of the ball's motion after leaving the student's hand until the ball reaches maximum height.

Air resistance has been ignored.

FIGURE 10

Velocity in m/s



Time in seconds



[Turn over]	
Maximum height =	_ m
Use FIGURE 10, on the opposite page. [2 marks]	
Calculate the maximum height reached by the ball.	
The maximum height is equal to the area between the line and the horizontal axis.	9
07.1	



0	7	2

Calculate the gradient of the line in FIGURE 10, on page 50.

Use the equation:

gradient =	change in y value change in x value	[2 marks]	
Gradient =	=		

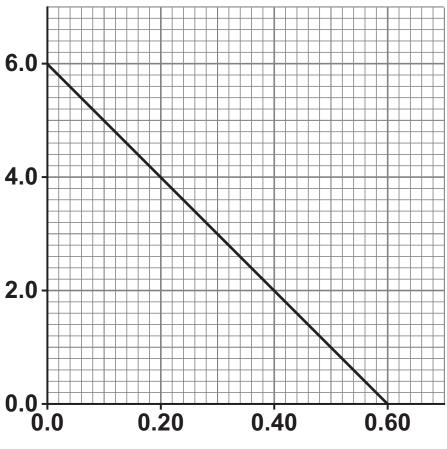


07.3
What does the gradient of the line in FIGURE 10, on page 50, represent? [1 mark]
Tick (✓) ONE box.
The deceleration of the ball.
The distance travelled by the ball.
The speed of the ball.
[Turn over]



FIGURE 10 is repeated here.

Velocity in m/s



Time in seconds



07.4
In FIGURE 10 air resistance was ignored.
What would happen to the motion of the ball in FIGURE 10 if air resistance was included? [2 marks]
Tick (✓) TWO boxes.
The deceleration would be greater.
The final speed would be greater.
The initial kinetic energy would be less.
The initial velocity would be less.
The maximum height of the ball would be less.
[Turn over]



The student threw a second ball vertically upwards into the air.

The maximum height reached by the second ball was 5.0 m.

The student caught the ball at the same height that the ball was thrown from.

The displacement of the ball is the straight-line distance between the start height and the end height.

0 7.5 What is the total distance the ball travels?	[1 mark]
Tick (✓) ONE box.	
0.0 m	
5.0 m	



10.0 m

0 7.6 What is the displacement of the ball when the student catches the ball? [1 mark]	
Tick (✓) ONE box.	
0.0 m	
5.0 m	_
10.0 m	-



0	8

This question is about hydrocarbon fuels.

The COMPLETE combustion of a hydrocarbon fuel produces carbon dioxide and one other product.

0 8.1

Name the other product of the COMPLETE combustion of a hydrocarbon fuel.

Do NOT refer to carbon dioxide. [1 mark]



08.2
Describe the test for carbon dioxide.
Give the result if carbon dioxide is present. [2 marks]
Test
Result

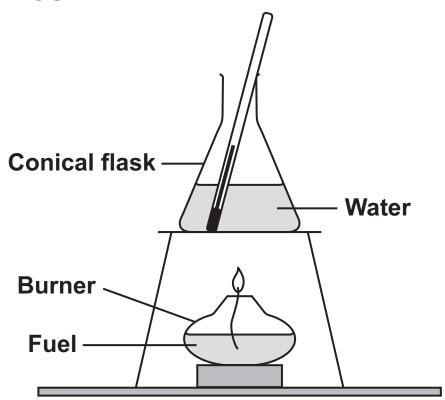


The combustion of hydrocarbon fuels releases energy.

A student investigated the energy released by three different fuels A, B and C.

FIGURE 11 shows the apparatus.

FIGURE 11



This is part of the method used.

- 1. Put fuel A in the burner.
- 2. Pour 100 cm³ of water into the conical flask.
- 3. Record the temperature of the water.
- 4. Light the fuel and heat the water for 5 minutes.
- 5. Record the final temperature of the water.
- 6. Repeat steps 1 to 5 using fuel B and then using fuel C.



0 8 . 3

The student also determined the mass of fuel burnt.

Describe how the student could determine the n fuel burnt. [2 marks]	nass of
0 8 . 4 Give ONE control variable in the investigation.	[1 mark]



0 8 . 5

TABLE 5 shows the results.

TABLE 5

FUEL	MASS OF FUEL BURNT IN GRAMS	TEMPERATURE INCREASE OF WATER IN °C
Α	1.72	40
В	1.65	45
С	1.23	50

most energy per gram of fuel. [2 marks]			

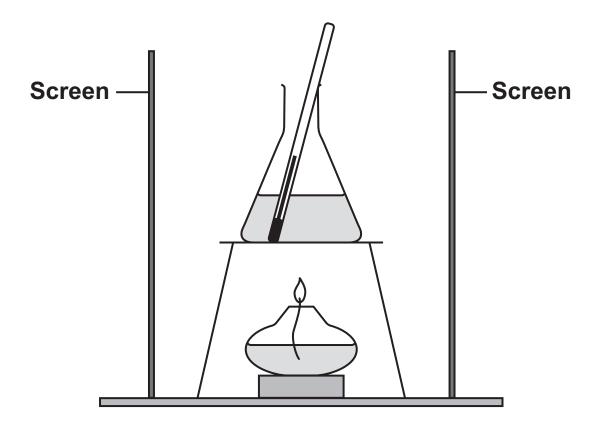


0 8 . 6

The student repeated the investigation with screens around the apparatus.

FIGURE 12 shows the apparatus with screens.

FIGURE 12



Give ONE reason why putting screens around the apparatus could improve the accuracy of the investigation. [1 mark]



0 8 . 7	
How would stirring the water improve the accuracy of the investigation? [1 mark]	
	- -
	10



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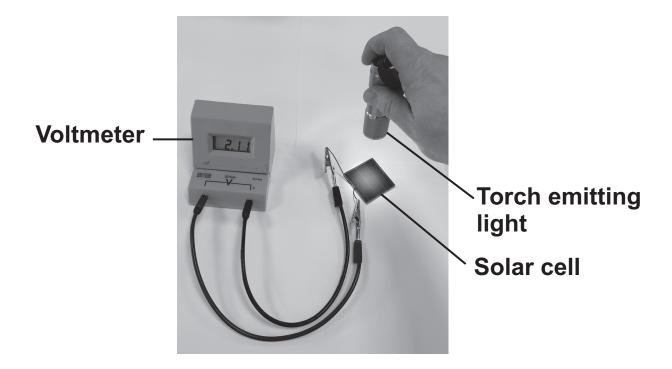
0 9

A solar cell generates a potential difference when light shines on its surface.

A student investigated how the potential difference varied with the light intensity at the surface of the solar cell.

FIGURE 13 shows some of the equipment used by the student.

FIGURE 13



The voltmeter displayed the potential difference generated by the solar cell.

The student varied the light intensity by changing the height of the torch above the solar cell.



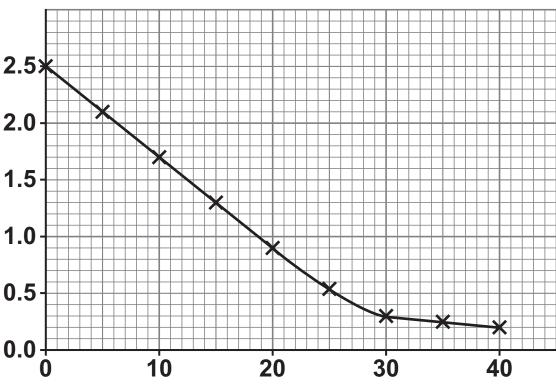
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FIGURE 14 shows the results.

FIGURE 14

Potential difference in volts



Height of torch above solar cell in cm

0 9		1
-----	--	---

Describe a method the student could have used to obtain the results shown in FIGURE 14. [6 marks]





The circuit symbol for a solar cell is:

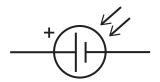
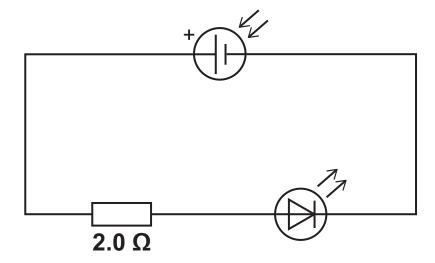


FIGURE 15 shows the solar cell in a circuit with an LED and a resistor.

FIGURE 15





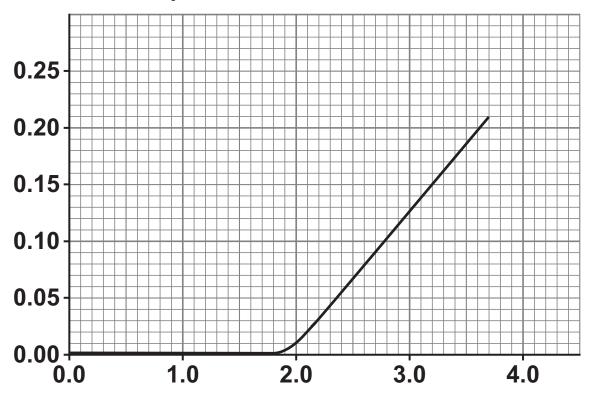
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FIGURE 16 shows how the current in the LED varies with the potential difference across the LED.

FIGURE 16

Current in amps



Potential difference in volts



0 9 . 2	0	9		2
---------	---	---	--	---

What is the range of potential difference values for which the LED emits light in FIGURE 16, on the opposite page? [1 mark]

Use the Physics Equations Sheet to answer questions 09.3 and 09.4.

0 9.3

Which equation links current (I), potential difference (V) and resistance (R)? [1 mark]

Tick (✓) ONE box.



$$V = I^2R$$

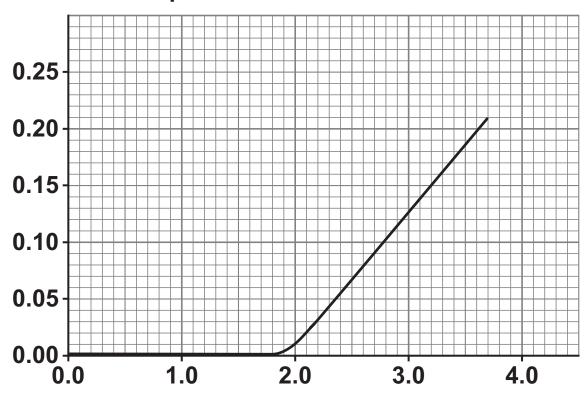
$$V = IR^2$$

$$V = \frac{I}{R}$$



FIGURE 16 is repeated here.

Current in amps



Potential difference in volts

0 9 . 4

Determine the resistance of the LED when the potential difference across the LED is 2.7 V.

Use FIGURE 16. [4 marks]



Resistance =	
0 9 . 5	
Describe how the resistance of the LED varies as the p.d. increases from 0 V to 3.7 V.	
Use data from FIGURE 16. [2 marks]	



1 0

FIGURE 17 shows two different designs of wind turbine.

FIGURE 17

THREE-BLADE WIND TURBINE



BLADELESS WIND TURBINE



To generate electricity, the three-blade wind turbine rotates about an axis.

To generate electricity, the bladeless wind turbine oscillates from side to side.



TABLE 6 gives information about the two designs.

TABLE 6

FEATURE	THREE-BLADE TURBINE	BLADELESS TURBINE
Lubrication needed	Yes	No
Maintenance costs	High	Low
Noise level	High	Low
Power output	High	Low
Risk to flying birds	Yes	No

1	0	1

Feature

Which feature of the three-blade turbine is an advantage compared with the bladeless turbine?

Use TABLE 6.

Give a reason for your answer. [2 marks]

Reason



To generate electricity, the bladeless wind turbine oscillates from side to side.

FIGURE 18 shows the direction of the oscillations of the bladeless wind turbine.

FIGURE 18

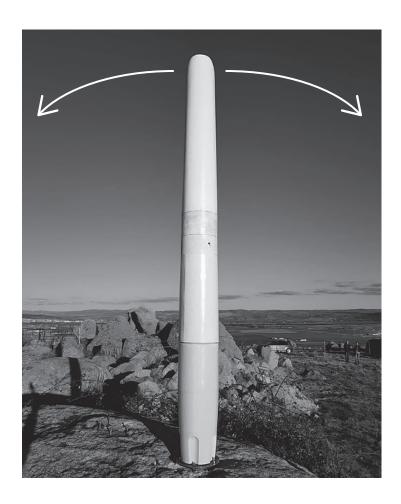
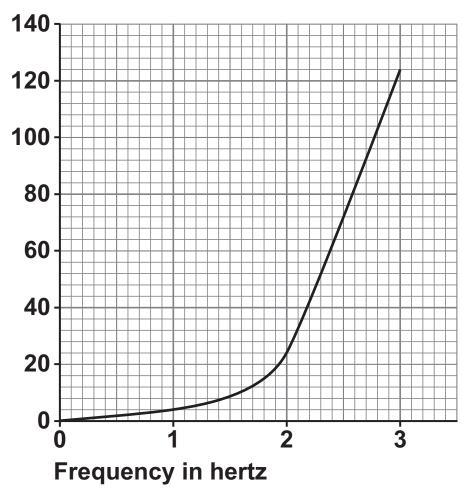




FIGURE 19 shows how the power output of the bladeless turbine varies with the frequency of the oscillation.

FIGURE 19

Power output in watts





1 0 . 2 Describe how the power output of the bladeless turbine varies with frequency. [2 marks]			



The energy from wind turbines can be used to recharge a battery.

Use the Physics Equations Sheet to answer questions 10.3 and 10.4.



Which equation links charge flow (Q), current (I) and time (t)? [1 mark]

Tick (✓) ONE box.

$$Q = I t^2$$

$$Q = I^2 t$$



10.4	
To fully recharge the battery, a charge of 216 000 C needs to flow through the battery.	
The current in the battery is 5.0 A.	
Calculate the time taken to fully recharge the battery.	
Give the unit. [4 marks]	
Time taken =	

END OF QUESTIONS

Unit _____



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Additional page, if required. Write the question numbers in the left-hand margin.



Additional page, if required. Write the question numbers in the left-hand margin.



Additional page, if required. Write the question numbers in the left-hand margin.





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For Examiner's Use			
Question	Mark		
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